

# C-H amination reactions via radical pathway; repurposing Hofmann-Löffler-Freytag reaction.

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LIGHT  RING



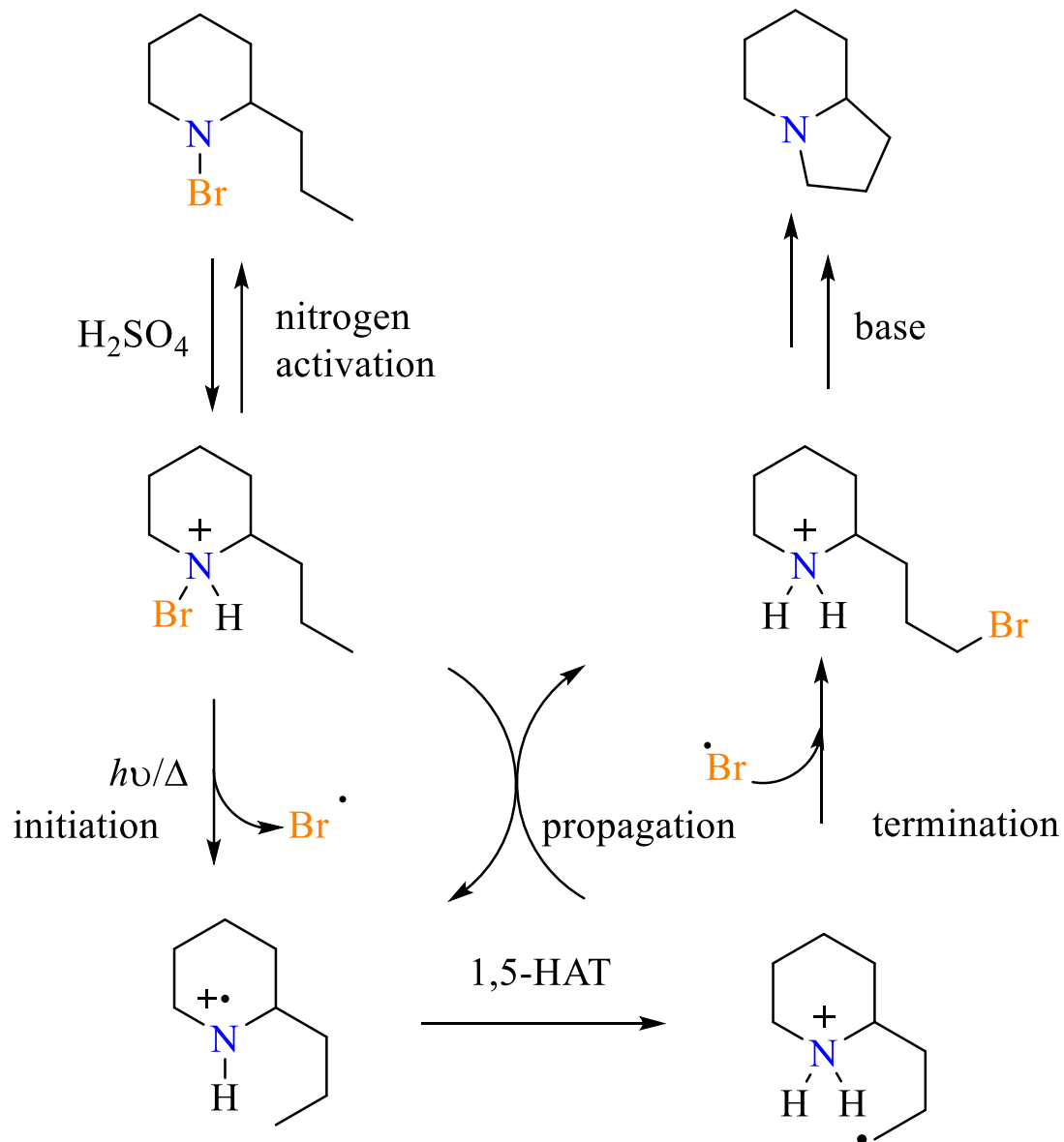
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# Hofmann-Löffler-Freytag reaction

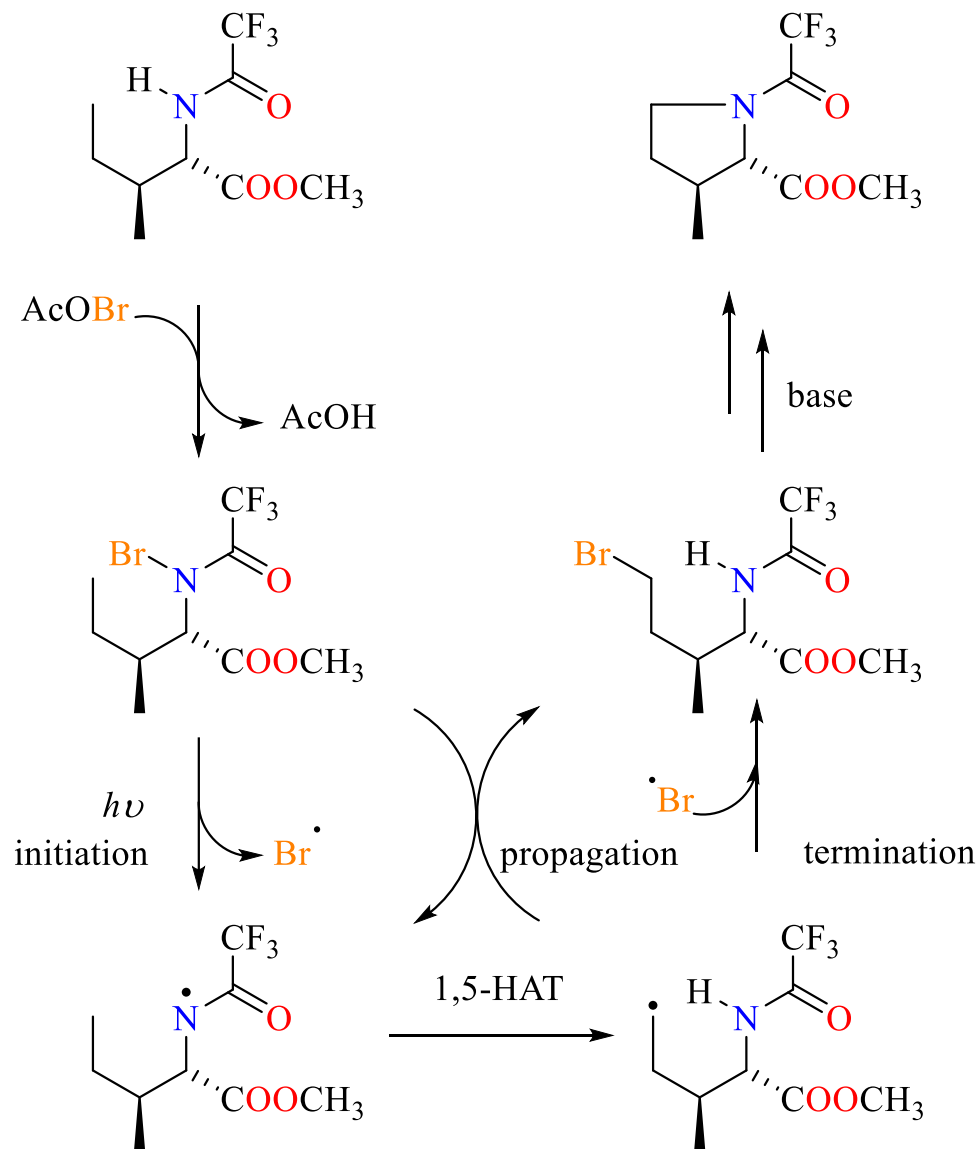
- Tool inside the C-H functionalization toolbox
- Product of the reaction is pyrrolidine and more sparsely piperidine
- It is defined as a late stage functionalization synthetic strategy
- Has the potential to become an organo-catalysed reaction in mild conditions in keeping with green chemistry principles

# Original Hofmann-Löffler-Freytag (HLF) reaction

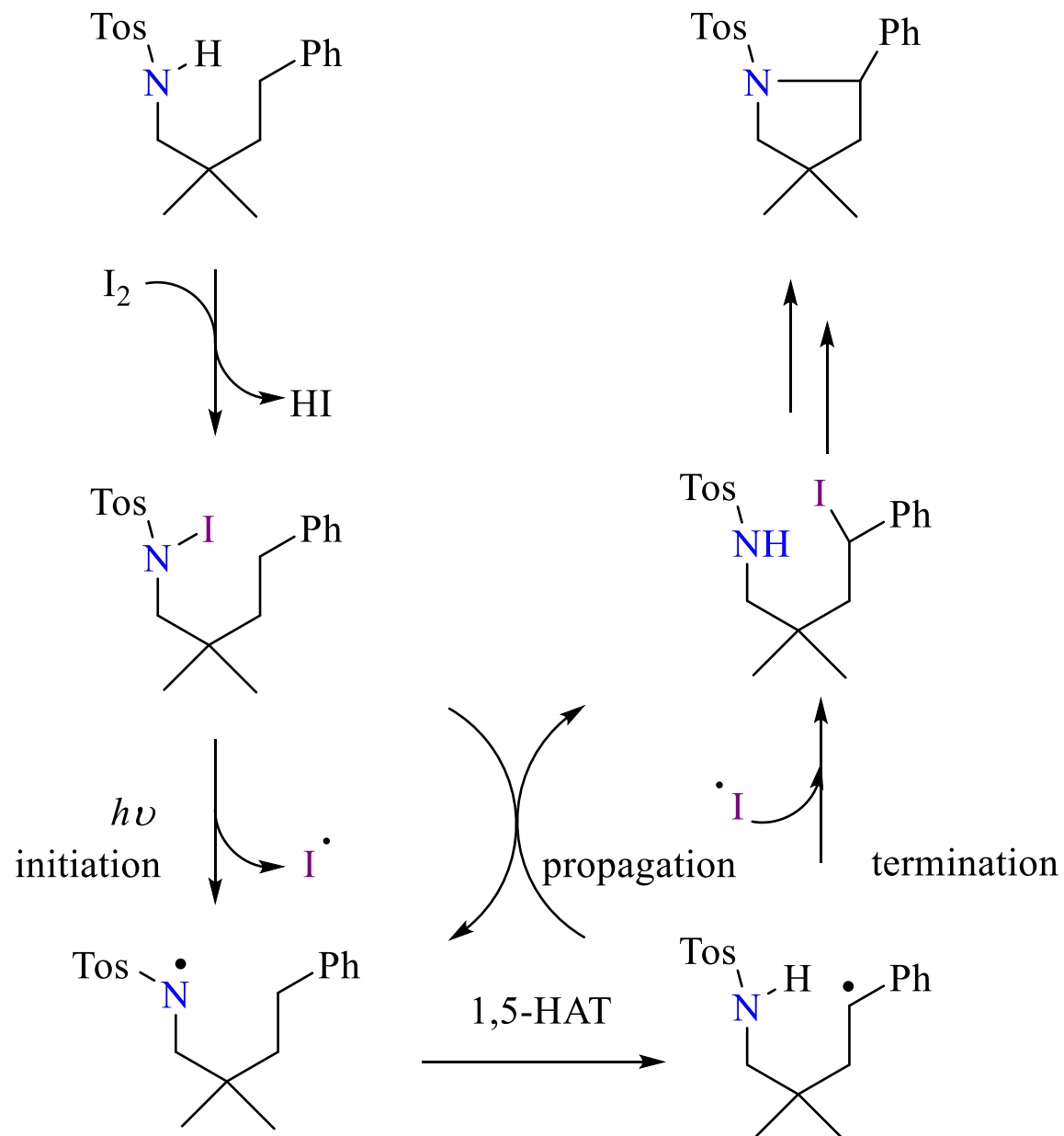


a) Hofmann, A. W. *Ber. Dtsch. Chem. Ges.* **1881**, 14, 2725; b) Hofmann, A. W. *Ber. Dtsch. Chem. Ges.* **1883**, 16, 558; c) Hofmann, A. W. *Ber. Dtsch. Chem. Ges.* **1885**, 18 (5), 109 d) Löffler, K., Freytag, C. *Ber. Dtsch. Chem. Ges.* **1909**, 42, 3427

# "Corey modification" of the HLF procedure



# "Suarez modification" of the HLF reaction

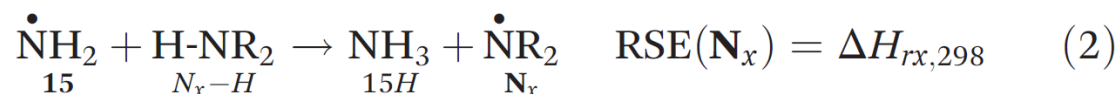


VIP Very Important Publication

## Radical Stability as a Guideline in C–H Amination Reactions

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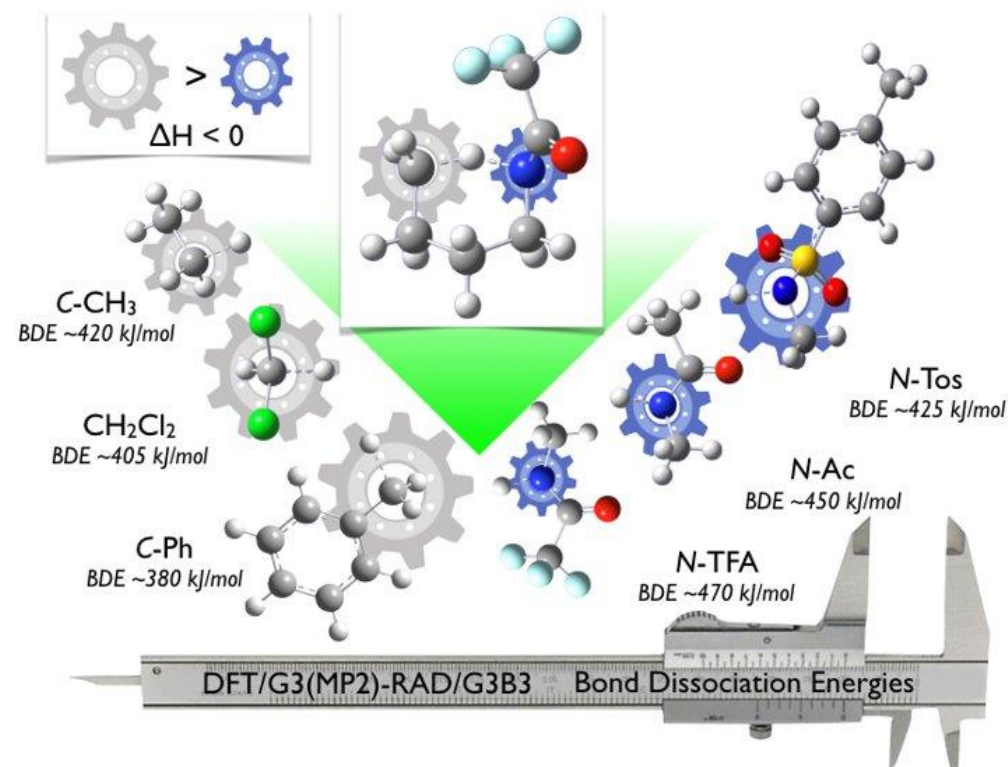
Fax: (+49)-89-2180-77738; phone: (+49)-89-2180-77737; e-mail: zipse@cup.uni-muenchen.de



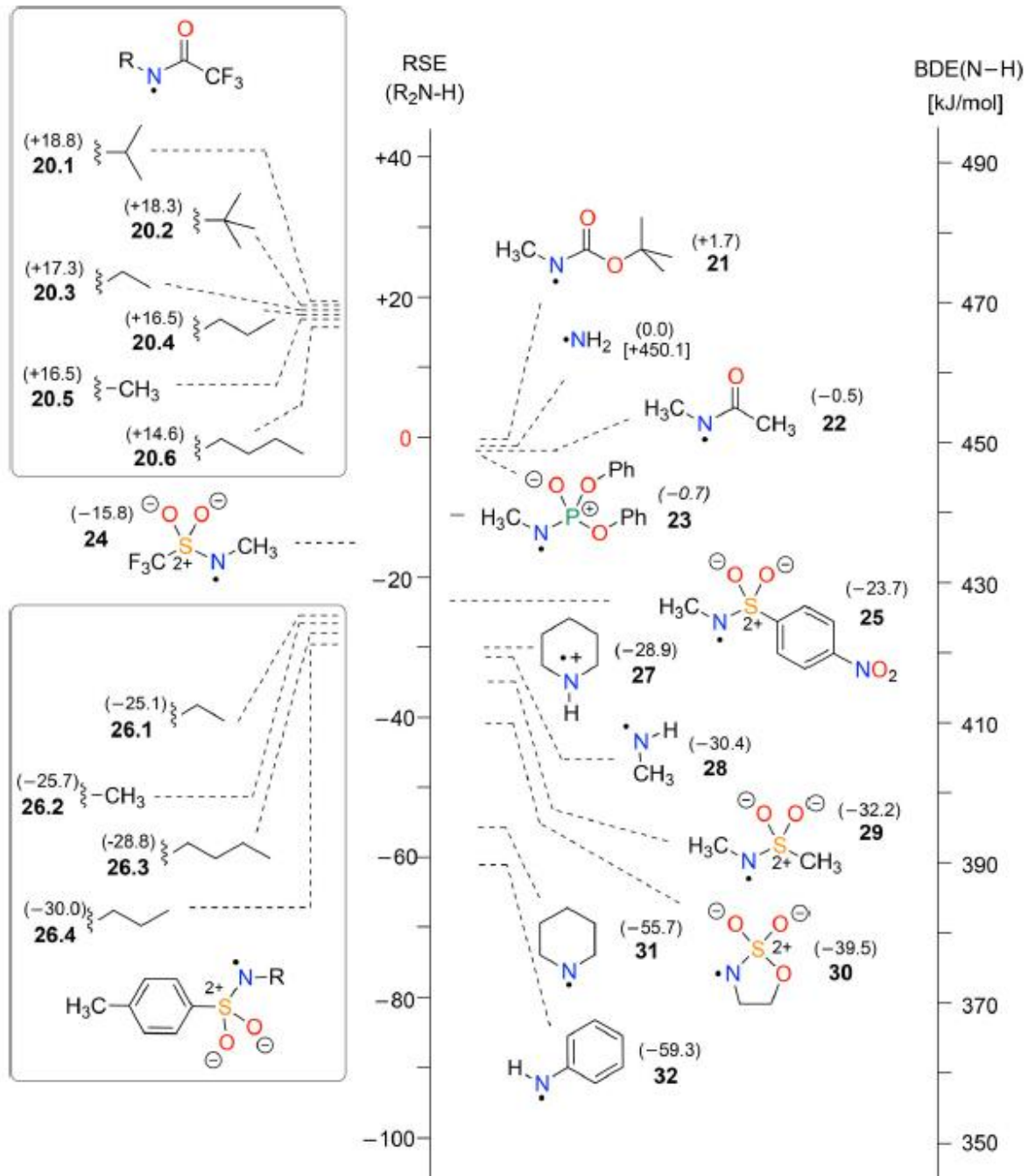
$$\text{BDE}(\text{C}_{xx}\text{-H}) = \text{BDE}(\text{1H}/\text{1}) + \text{RSE}(\text{C}_x) \quad (3)$$

$$\text{BDE}(\text{N}_x\text{-H}) = \text{BDE}(\text{15H}/\text{15}) + \text{RSE}(\text{N}_x) \quad (4)$$

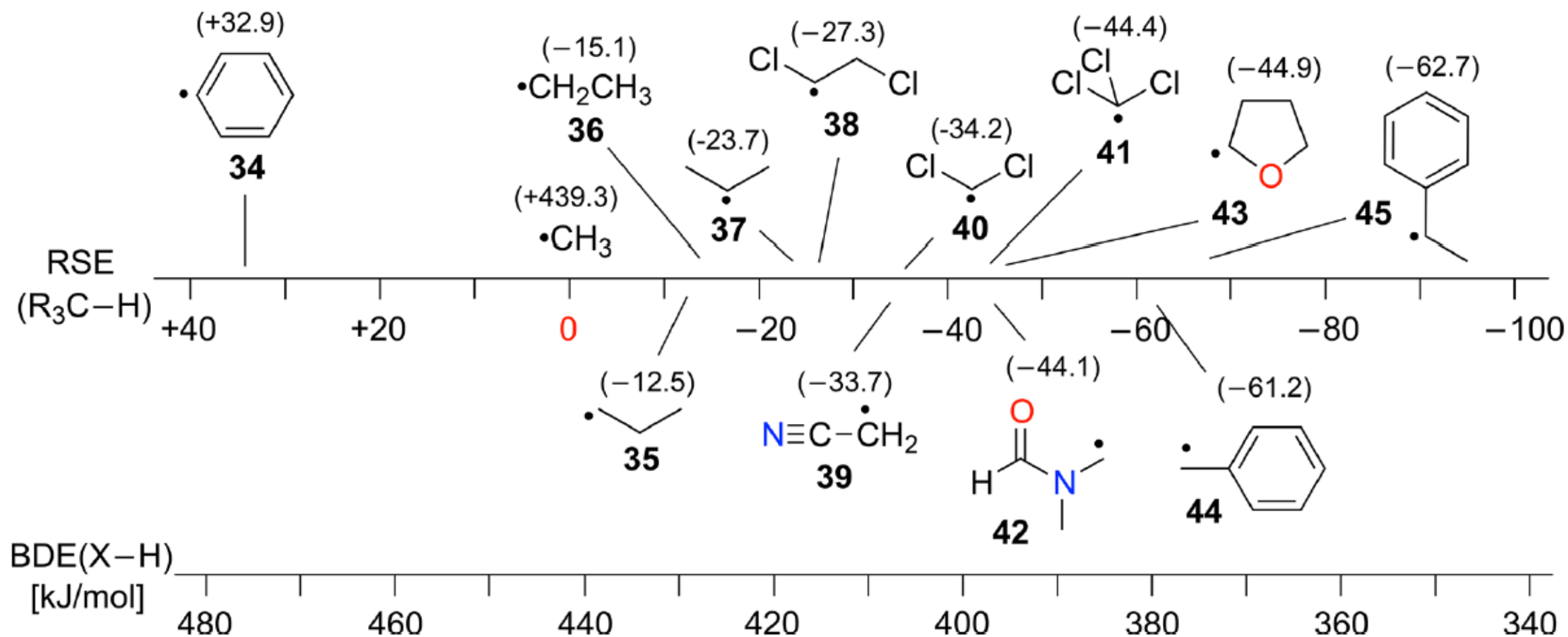
$$\Delta H_{298}^\ddagger = E_0 + \alpha \Delta H_{rxn,298} \quad (5)$$



# Radical stabilization energies (RSE) and bond dissociation energies (BDE) for aminyl radicals

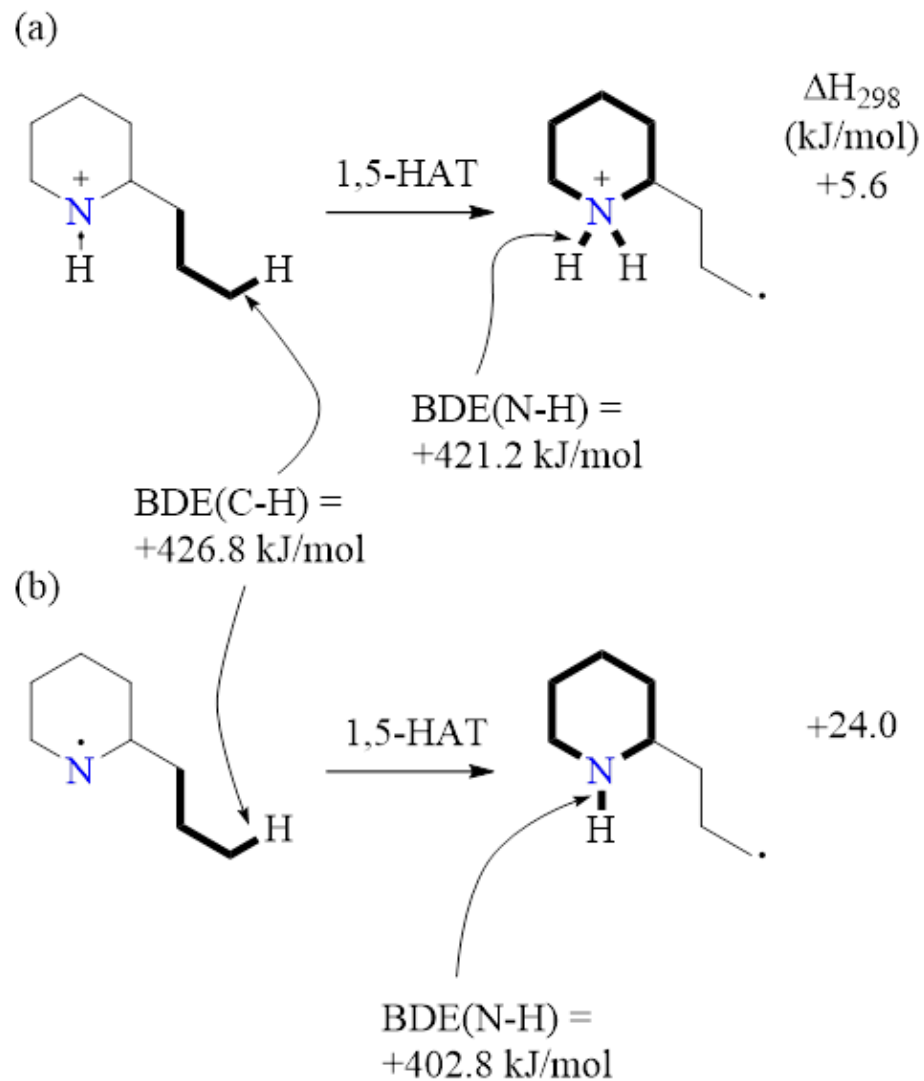


# RSE and BDE values for selected C-centred radicals

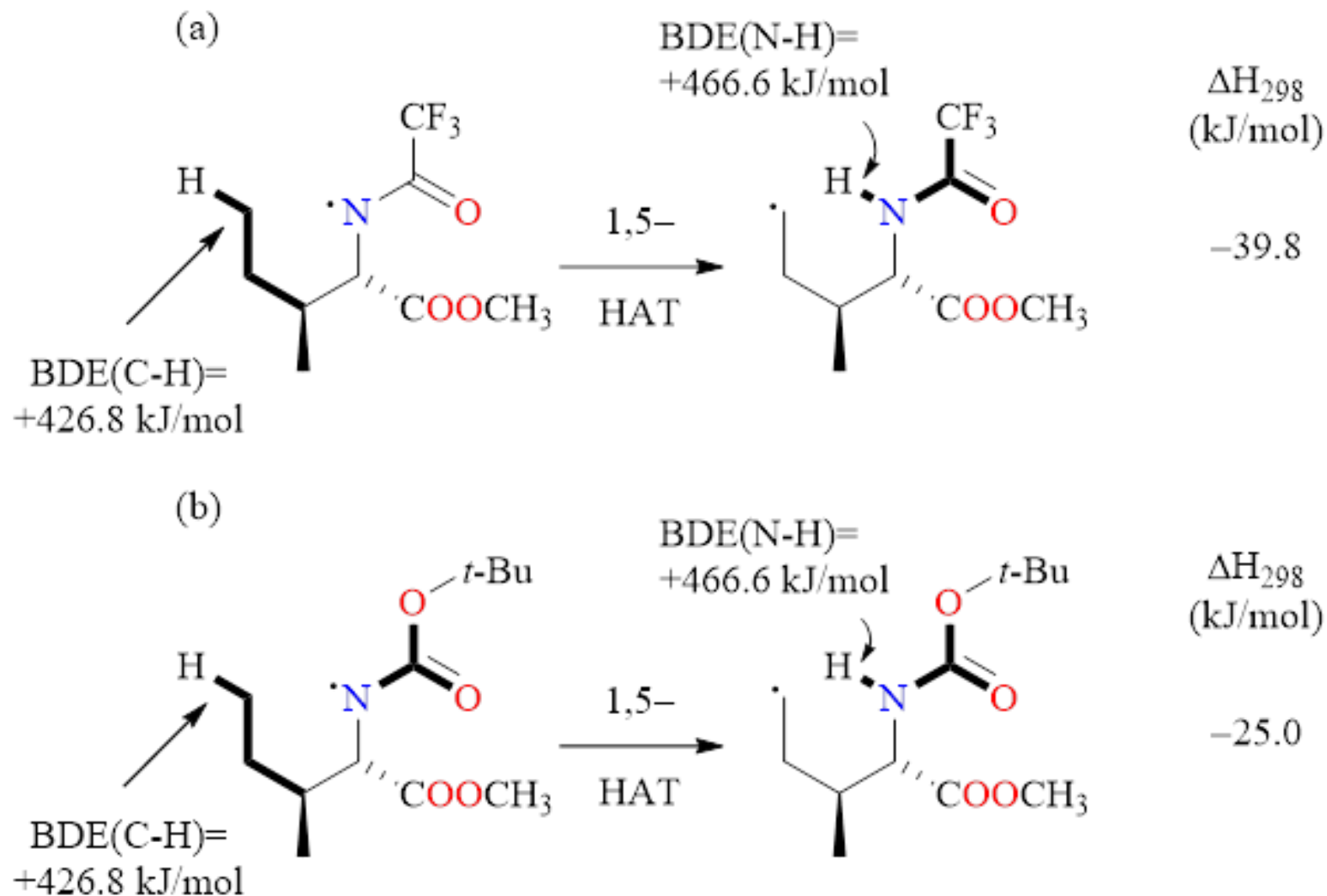




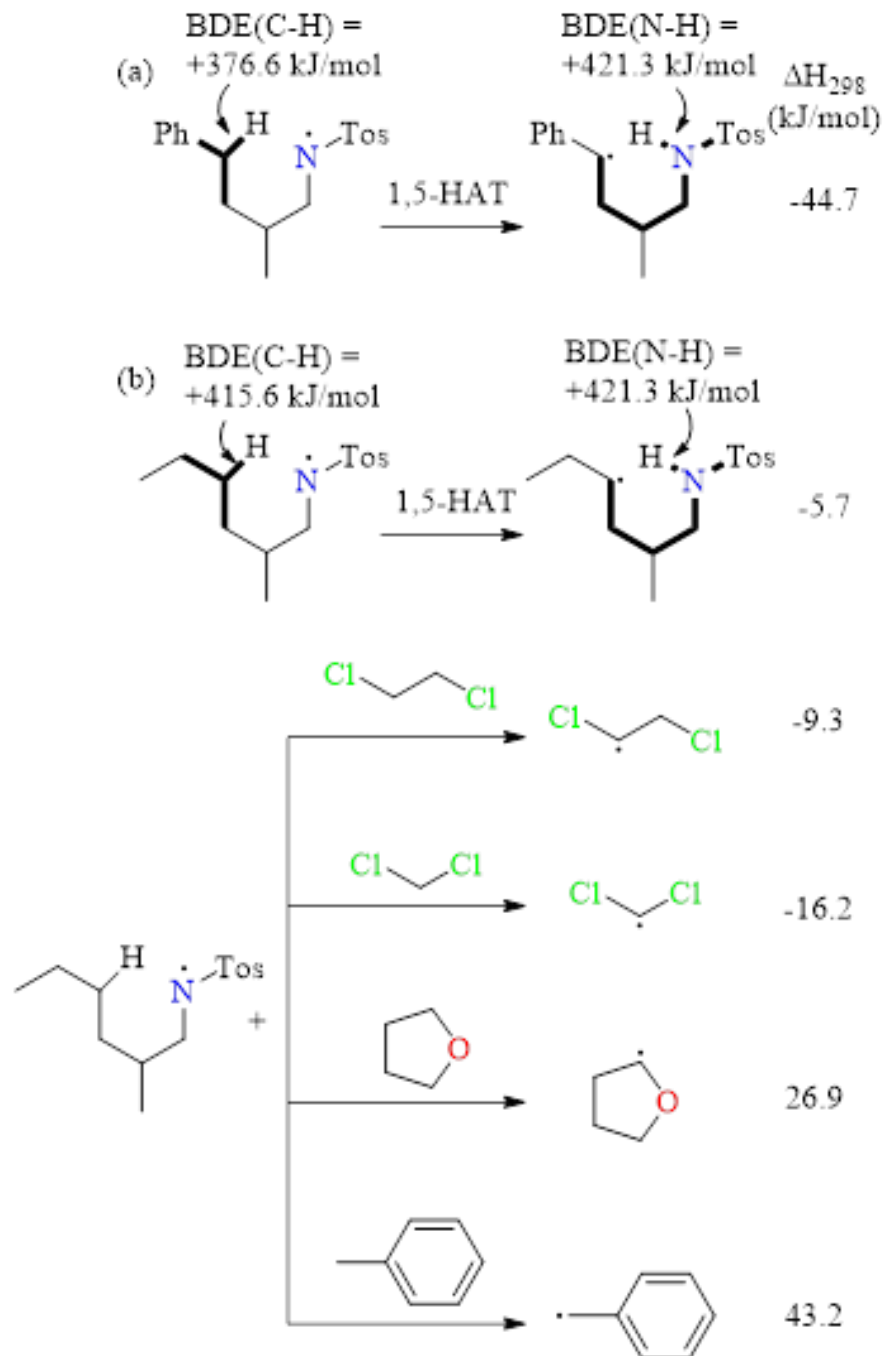
# The effect of reaction medium on the thermochemical profile of the HAT step for the classical HLF reaction scheme



# Thermochemical profile of 1,5-HAT step for the "Corey modification"



# Thermochemical profile of 1,5-HAT step within the "Suarez modification" and intermolecular HAT involving solvent molecules

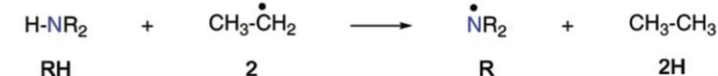


## Role of substituents in the Hofmann–Löffler–Freitag reaction. A quantum-chemical case study on nicotine synthesis†‡

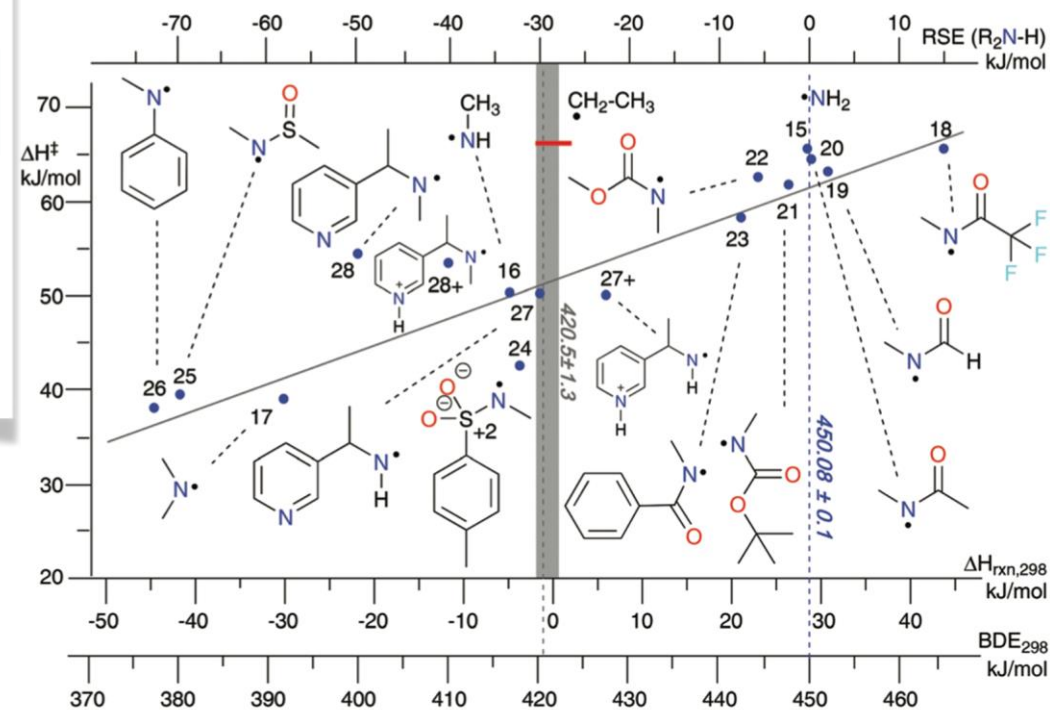
Sofia Shkunnikova,<sup>a</sup> Hendrik Zipse<sup>lb</sup> and Davor Šakić<sup>ld</sup>\*<sup>a</sup>

 Check for updates

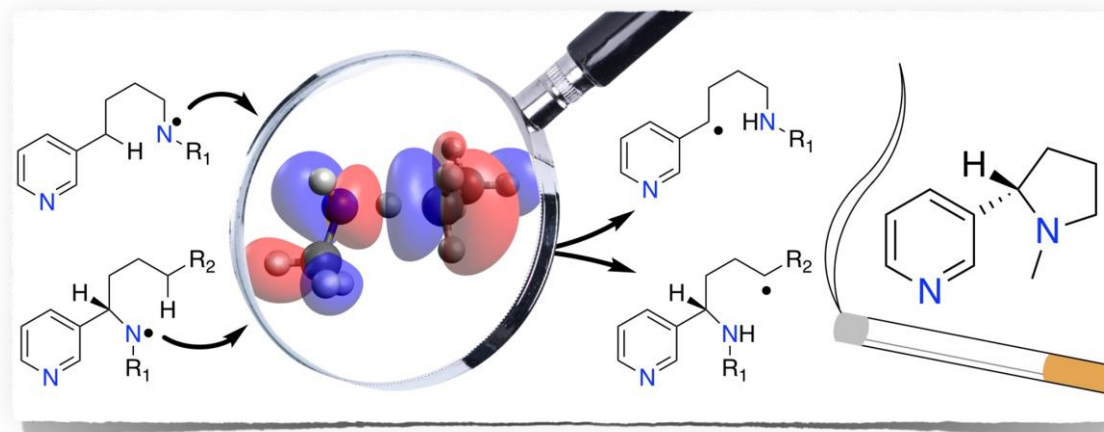
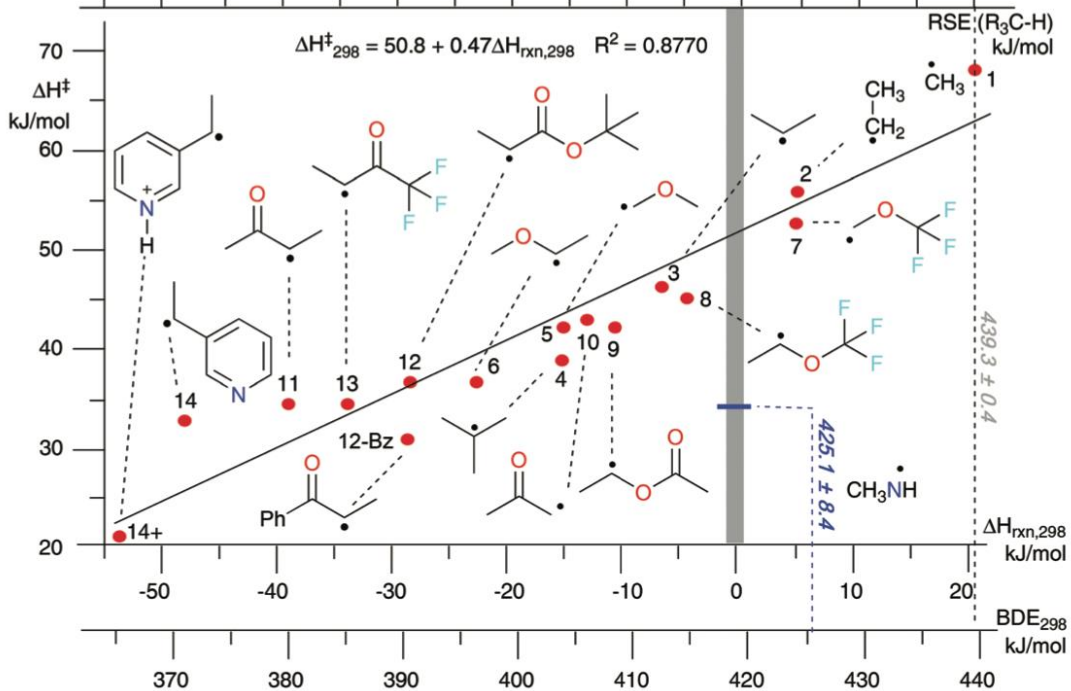
Cite this: *Org. Biomol. Chem.*, 2021, 19, 854



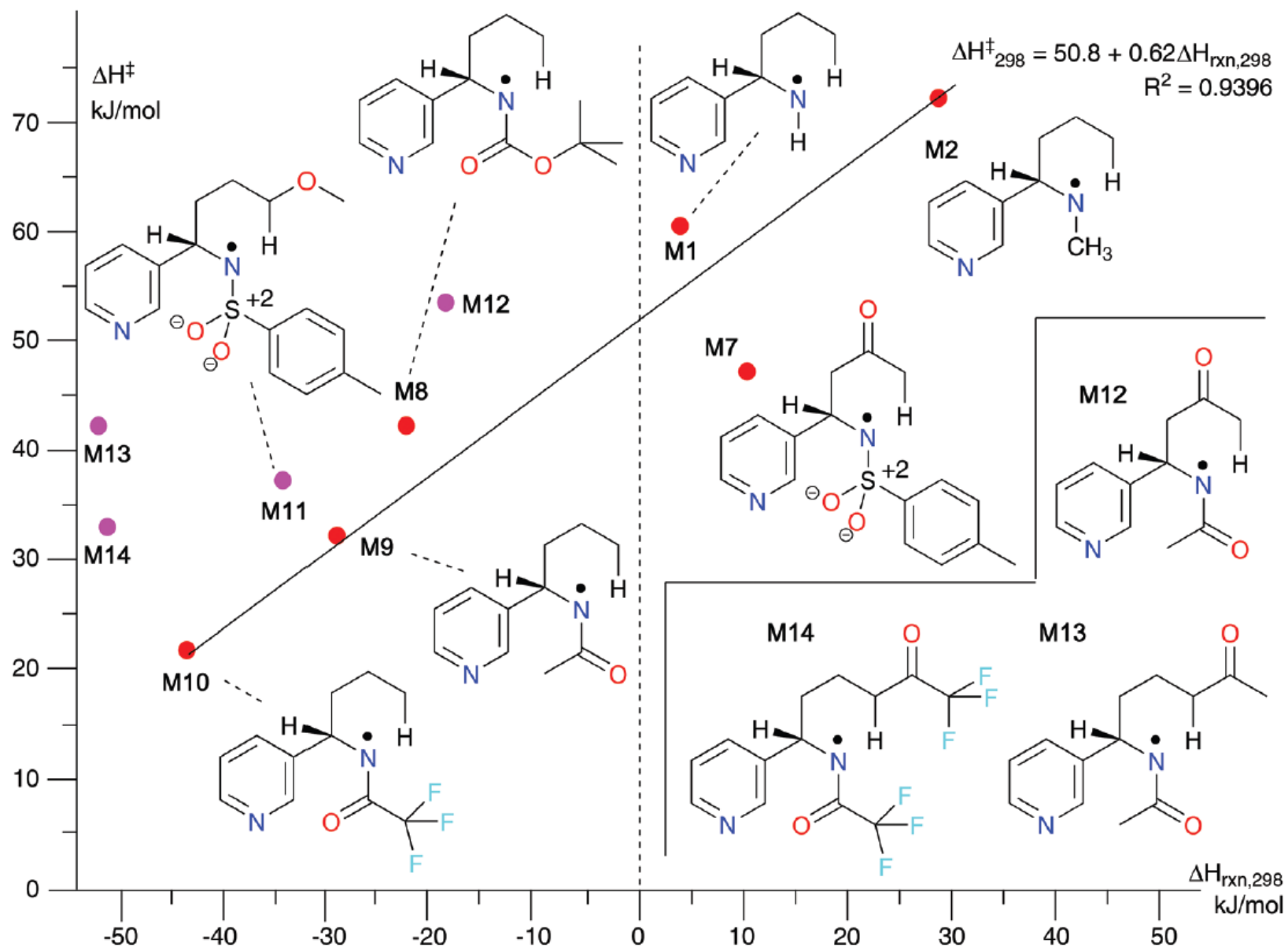
$$\Delta H^\ddagger_{298} = 51.6 + 0.36\Delta H_{\text{rxn},298} \quad R^2 = 0.9173$$



$$\Delta H^\ddagger_{298} = 50.8 + 0.47\Delta H_{\text{rxn},298} \quad R^2 = 0.8770$$



# Application of Bell-Evans-Polany principle



# Summary

- Calculations have been done with a goal to elucidate the parameters guiding the HLF reaction.
- However, much work is left to be done!
- A switch that accounts for 1,5 or 1,6-HAT regioselectivity is to be determined.
- The role of the oxidant in the shift from inter- to intramolecular HAT steps has to be quantified.
- A switch between ionic and radical pathways is to be established.

# Thank you for you attention!



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